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Effectiveness of soaps

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and detergents as  
cleansing agents

Advantages of soaps

1. Soaps are effective  
cleansing agents in soft  
water, that is water does  
not contain  $Mg^{2+}$  and  
 $Ca^{2+}$  ions. 2. Soaps do  
not cause pollution  
problems to the  
environment. This is  
because soaps are made  
from chemical found in  
animals and plants.

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5.1 Soap and Detergent.

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### SOAP AND DETERGENT

Definition: Soap A

sodium or potassium salt  
of long-chain fatty acid

General formula of soap:

$\text{RCOO-Na}^+$  or  $\text{RCOO-}$

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K+. 3. Detergent A

sodium or potassium salt  
of alkyl sulphonic acids  
or alkylbenzene

sulphonic acid. 4.

## 5.1 Soap and Detergent - SlideShare

Difference Between Soap  
and Detergent: Soaps:  
Detergents: Consist of a  
' -COONa ' group  
attached to a fatty acid  
having a long alkyl chain.

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Consist of a 'SO<sub>3</sub>Na' group attached to a long alkyl chain. They are not effective in hard water and saline water: They do not lose their effectiveness in hard water and saline water.

Soap and Detergent -  
Preparation, Difference  
between Soap ...

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Soap: Detergents: Made from: Natural materials found in animals and

plants: Synthetic materials from

petroleum: Formation:

Reaction between an alkali and a fatty acid:

Reaction between an alkali and a sulphuric

acid: Effectiveness in

hard water: No effective in hard water due to

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formation of scum.

Effective in hard water.

Do not form scum in  
hard water

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Taiwan Circling Swirl  
Variation, Cold Process  
Soap Making,  
(Technique Video #5) -

Duration: 13:42. Tree

Marie Soapworks

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## Chapter 5 Soaps And Chemistry Form 5 - Soap And Detergent (Bahasa Sarawak)

These include chemical etching (using vapor or liquid form of highly toxic HF, HNO<sub>3</sub>, and H<sub>2</sub>SO<sub>4</sub>), electrocleaning (by NaOH), solvent cleaning (by acetone, methanol, isopropanol, chlorinated, and

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chlorofluorocarbon  
solvents), soap/detergent  
cleaning, ultrasound  
cleaning, oxidative  
cleaning (UV/ozone),  
reactive plasma cleaning,  
and plasma etching  
(fluorine-based gases)  
(4). Toxic liquids, fumes,  
radiation, and other  
hazards involved in  
cleaning processes lead to  
potential health and  
safety ...

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Soaps (Detergent) - an  
overview | ScienceDirect  
Topics

The basic structure of the detergent molecule is similar in nature to a soaps. It will dissolve in water to produce sodium ion ( $\text{Na}^+$ ) and detergent ions. A detergent ion consists of two parts which are known as the hydrophilic part and the

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hydrophilics part.

## Detergent - SPM Chemistry

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The soap is ready when  
the soap solution obtains  
very thick gel-paste like  
constituency. The soaps  
are alkaline with pH  
varying between 8 and

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11. It is easier with the pH strips as they show the exact stage of soap formation and completion of the reaction between sodium hydroxide and oils.

Soaps and Detergents |  
Classification and  
Application of ...

The Car Wash  
Detergents & Soaps  
market, study now

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available at Ample Market Research, is a detailed sketch of the business sphere in terms of current and future trends driving the profit matrix. The report also indicates a point-wise outline of market share, market size, industry partakers, and regional landscape along with statistics, diagrams, & charts elucidating various

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noteworthy ...

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Car Wash Detergents &  
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Soaps Market: Poised  
For a Strong ...

Cleansing Action of  
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Most of the dirt is oily in  
nature and oil does not  
dissolve in water. The  
molecule of soap  
constitutes sodium or  
potassium salts of long-  
chain carboxylic acids. In

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the case of soaps, the carbon chain dissolves in oil and the ionic end dissolves in water. Thus, the soap molecules form structures called micelles. In micelles, one end is towards the oil droplet and the other end which is the ionic faces outside.

Cleansing Action Of  
Soaps And Detergents -  
Micelle ...

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Gradually, soapmaker ' s guilds appeared, with Italy, Spain, France and England becoming centers of soap manufacture. King James granted a monopoly to a soapmaker in England in 1622 for \$100 000 a year (The Soap and Detergent Association, 1981). From that time forward, the desire for and use of soap has increased

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tremendously into  
modern times.

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