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Chapter 6.1 Slides
6.1 Organizing the Elements >Today ' s Periodic Table 21 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. . Mendeleev developed his table before scientists knew about the structure of atoms. • He didn ' t know that the atoms of each element contain a unique number of protons. • Recall that the number of protons is the

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SECTION 6.1 ORGANIZING THE ELEMENTS (pages 155 – 160) This section describes the development of the periodic table and explains the periodic law. It also describes the classification of elements into metals, nonmetals, and metalloids. Searching For An Organizing Principle (page 155) 1.

SECTION 6.1 ORGANIZING THE ELEMENTS (pages 155 – 160) (page 155)
SECTION 6.1 ORGANIZING THE ELEMENTS 1. Which element listed below should have chemical properties similar to fluorine (F)? a. Li b. Si c. Br d. Ne 2. Identify each element as a metal, metalloid, or nonmetal. a. fluorine b. germanium c. zinc d. phosphorus e. lithium 3. Which of the following is not a transition metal? a. magnesium b. titanium c. chromium d. mercury 4.

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