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Kids will learn about some important
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Atomic Structure Worksheet List the 3
subatomic particles, their charges, masses
and where in the atom they are located in
the chart below: Describe how to determine
the number of each subatomic particle in an
atom below Fill in the chart for the elements
below. Round all atomic masses to the
nearest whole number: Draw an atom below
and label

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The protons and neutrons are at the center of the atom, the nucleus, with empty space around it.

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*Here ' s how to draw an atom of any element using the Bohr model of the atom: Protons and neutrons go in the nucleus (center). Electrons go in energy levels of the electron cloud.

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~~5 — Structure of an atom and isotopes~~

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A. Structure of an atom 1. Fill in the blanks concerning sub-atomic particles. (a) There are three sub-atomic particles in atoms. They are _ protons _, _ neutrons _ and _ electrons _.

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tells you the number of _____ in a neutral atom of that element. The atomic number gives the “ identity ” of an element as well as its location on the periodic table. No two different elements will have the _____ atomic number. 4. The _____ of an element is the average mass of an element ' s naturally occurring atom, or

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Name three sub atomic particles of an atom.
2. Define atomic number and mass number.
3. Describe the particle scattering experiment conducted by Rutherford with a diagram. 4.

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Students define an atom, label the parts, describe and compare the protons, neutrons, and electrons, and determine the meaning of atomic number and mass number. This clear and c. Subjects:

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Worksheet for chapter: Structure of the Atom is presented below. The worksheets are provided for practice and self evaluation of students of class 9. Solutions are provided at the end of the page. (1) Write electronic configuration of element X which has atomic number = 15. Write its valency also.

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General Chemistry Mr. MacGillivray
Atomic Structure Worksheet 1. The number of protons in the nucleus of an atom of a particular element is called the _____

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_____ of that element. 2. In a neutral element, the number of protons must equal the number of _____. 3.

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